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reactions, their examples and corresponding chemical equations. A chemical reaction is a process in which new substances with new properties are formed. It involves rearrangement of atoms and breaking of chemical bonds ...

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Chemical reactions are represented by balanced chemical equations, with chemical formulas symbolizing reactants and products while chemical equations give information in two major areas: first, they tell us what substances are reacting (those being used up) and what substances are products (those being

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made). Second, the coefficients of a balanced equation tell us in what ratio the substances ...

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Hence, equations III and VII are the equations of rate constants of zero and first order reactions respectively. We can find rate constants, initial and final concentrations and the time taken for the reaction to occur using these reactions. A Solved Question for You. Q: The order of a reaction is:

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